Synthesis, Growth and Characterization of Magnesium Sulfate Doped L-Leucinium Oxalate Semi-Organic Crystal for Laser Applications

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Abstract. Semi-Organic Magnesium Sulfate-doped L-Leucinium oxalate (MSLO) single crystal was grown using slow evaporation method. Structure of the crystal was determined by single-crystal X-Ray Diffraction (XRD). Lattice parameters of MSLO crystal were determined. Optical transmittance study was carried out to find various linear optical parameters like band gap, reflectance, refractive index, absorption coefficient, extinction coefficient, optical conductivity. Vickers Microhardness study was carried out to find mechanical properties such as hardness, work hardening coefficient, brittleness index and fracture toughness of the grown MSLO crystal. Kurtz-Perry powder technique with Nd:YAG laser was applied to study the second harmonic generation (SHG) competency. By using Z-scan technique, third-order NLO efficiency of MSLO crystal was calculated. LDT study was also done for MSLO crystal. It is observed that magnesium sulfate-doped L-leucinium oxalate (MSLO) crystal has the superior properties compared to that of undoped L-leucinium oxalate crystal. EDS studies were conducted for MSLO crystal to check the elements present in the crystal.

KEY WORDS: Growth, XRD, optical conductivity, impedance, Microhardness, SHG, LDT.

1 Introduction

Non Linear Optical (NLO) crystals are being considered largely due to their widespread applications in optical communication, optical computing, optical information processing, laser technology and photonics [1, 2]. Non-centrosymmetric organic compounds have second-order nonlinear optical fea-
tures and they are superior to inorganics [3–5]. It is known that, amino acid complexes are found to be important NLO materials for the development of devices. Many amino acids reveal nonlinear optical properties, due to the presence of donor NH$_3^+$ and acceptor COO$^-$ as well as the chance of intermolecular charge transfer [6–8]. Many number of studies have been described on the characteristics of amino acid complexes and organic, inorganic and semi organic NLO crystalline samples have been grown and studied by different authors [8–10]. Anbuchezhiyan et al. have grown the crystal of undoped L-leucinium oxalate. It was characterized by XRD, CHN, FTIR, NMR, NLO, optical and hardness studies [11]. Bhaskaran et al. have carried out thermal, optical, electrical properties of L-leucinium oxalate crystal [12]. In this work, L-leucinium oxalate is considered as the base material and doping of an inorganic material like magnesium sulfate was done to alter various properties of the host crystal. The growth, single crystal XRD, optical, SHG, mechanical, impedance and laser damage threshold (LDT) properties of magnesium sulfate-doped L-leucinium oxalate (MSLO) crystals are explained in this communication.

2 Material and Methods

AR grade chemicals such as L-leucine, oxalic acid and magnesium sulfate with high purity were purchased. Slow evaporation crystal growth method has variety of benefits like low cost, easy growing, controlling the rate of evaporation and re-crystallization to improve the quality of crystal and hence in this work, this method was followed to grow magnesium sulfate doped L-leucinium oxalate

![Flow diagram for crystal growth process.](image-url)
(MSLO) crystal. L-leucine and oxalic acid in 1:1 molar ratio and 1 mole% of magnesium sulfate were taken and dissolved in double distilled water. Care was taken to consider the solution was a supersaturated solution. Solution was stirred using magnetic stirrer attached with hot plate for about 4 hours. Then the solution was filtered using good quality Whatmann filter papers. The filtered solution was taken in a beaker which is closed by a holed polythene paper. The solution was allowed for slow evaporation. The flow diagram for crystal growth process is presented in Figure 1. The same procedure was also followed for the preparation of pure L-leucinium oxalate crystal.

Initially, seed crystals were grown. They were further purified by re-crystallization process [13]. By placing good quality seed crystals in supersaturated solution, single crystal of MSLO crystals were grown after the growth period of 28 days. Schematic Diagram of the reaction mechanism is given in Figure 2. The grown MSLO crystal is depicted in Figure 3. It is observed that the grown MSLO crystal is colourless and transparent. The size of the obtained crystal is $18 \times 4 \times 4 \text{ mm}^3$.

3 Results and Discussion

3.1 Single crystal X-ray diffraction studies

X-Ray Diffraction (XRD) is the principle to find crystal structure of a crystalline material because it acts like a three-dimensional grating. Single crystal XRD data were measured on a Bruker SMART KAPPA APEX II CCD
single-crystal X-ray diffractometer using graphite monochromatized MoKα radiation (\(\lambda = 0.71073\ \text{Å}\)). Obtained lattice dimensions of MSLO crystal are \(a = 5.671(4)\ \text{Å}\), \(b = 9.758(3)\ \text{Å}\), \(c = 9.889(3)\ \text{Å}\), and \(\alpha = 87.54(3)^\circ\), \(\beta = 98.83(2)^\circ\), \(\gamma = 101.04(3)^\circ\) and \(V = 542.63(4)\ \text{Å}^3\). The data reveals that, MSLO crystal crystallizes in triclinic structure with space group P1. It is found that the crystal structure of undoped L-leucinium oxalate [14] is same as compared to that of MSLO crystal but the lattice parameters were slightly changed due to inclusion of magnesium sulfate in the interstitial positions of the lattice of host L-leucinium oxalate crystal. XRD data for MSLO crystal is given in Table 1.

Table 1. Single-crystal XRD data for MSLO crystal

<table>
<thead>
<tr>
<th>Identity code</th>
<th>MSLO</th>
</tr>
</thead>
<tbody>
<tr>
<td>Empirical formula</td>
<td>C(<em>6)H(</em>{14})NO(_2).C(_2)HO(_4).MgSO(_4)</td>
</tr>
<tr>
<td>Molecular weight</td>
<td>341.57 g/mol</td>
</tr>
<tr>
<td>Temperature</td>
<td>293(2) K</td>
</tr>
<tr>
<td>Wavelength</td>
<td>0.71073 Å</td>
</tr>
<tr>
<td>Crystal system</td>
<td>Triclinic</td>
</tr>
<tr>
<td>Space group</td>
<td>P1</td>
</tr>
<tr>
<td>Lattice parameters (Å)</td>
<td>(a = 5.671(4)\ \text{Å}, b = 9.758(3)\ \text{Å}, c = 9.889(3)\ \text{Å} )</td>
</tr>
<tr>
<td></td>
<td>(\alpha = 87.54(3)^\circ, \beta = 98.83(2)^\circ, \gamma = 101.04(3)^\circ)</td>
</tr>
<tr>
<td>Unit cell volume</td>
<td>(V = 542.63(4)\ \text{Å}^3)</td>
</tr>
</tbody>
</table>

3.2 UV-Vis spectral assessment

In UV-visible spectroscopy, moving of electrons from lower state to higher energy levels occurs and hence it gives information about the molecule structure. To record the transmittance spectrum of MSLO crystal Perkin Elmer make Lambda 35 UV-Visible Spectrometer with the specification Range 190 nm to 1100 nm was used. The obtained spectrum is depicted in Figure 4. The spectrum reveals that the grown MSLO crystal’s lower cut-off wavelength is 218 nm.

3.2.1 Band gap

Optical band gap \((E_g)\) of the sample was determined using the relation \(E_g = \frac{1240}{\lambda}\), where \(\lambda\) is the wavelength of light. The band gap value was determined as 5.69 eV and this value is observed to be high. This indicates that MSLO crystal is an insulating material. The purpose of doing UV-Vis spectral study is to find the transmittance, absorption coefficient, optical band gap, extinction coefficient, optical conductivity, complex dielectric constant and linear refractive index and suitability crystal for nonlinear optical (NLO) applications [15].
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3.2.2 Absorption coefficient

Absorption coefficient ($\alpha$) is estimated using expression

$$\alpha = \frac{2.303 \log (1/T)}{t},$$

where $T$ is the transmittance in decimal points and $t$ is the thickness of the crystal. Variation of absorption coefficient with wavelength for MSLO crystal is given in Figure 5.

![Figure 5. Plot of absorption coefficient versus wavelength of MSLO crystal.](image)
Tauc’s relation is given by

\[ \alpha = \frac{A(h\nu - E_g)^{1/2}}{h} \]

where \( E_g \) is the optical band gap, \( h \) is Planck’s constant, \( \nu \) is the frequency and \( A \) is a constant [16]. The plot of variation of \((\alpha h\nu)^2\) versus photon energy \( h\nu \) is presented in Figure 6. Optical band gap was evaluated by the extrapolation of the linear part to the X-axis and the value is found to be 5.70 eV and this value indicates that MSLO crystal is a dielectric material.

### 3.2.3 Extinction coefficient

Extinction coefficient is the amount of light energy lost per unit thickness in a given medium due to scattering and absorption. Extinction coefficient of MSLO crystal was calculated using the relation \( K = \frac{\alpha \lambda}{4\pi} \), where \( \alpha \) is the linear absorption coefficient and \( \lambda \) is the wavelength of light. Variation of extinction coefficient with wavelength is shown in Figure 7. From the figure, it is observed that, the extinction coefficient is high at fundamental absorption in the UV region. In the wavelength region 500–1000 nm, extinction coefficient increases slightly with increase of wavelength. From the result, it is found that, extinction coefficient of MSLO crystal is very low. It shows the grown crystal is a good optical material with very low energy loss.
3.2.4 Reflectance

Reflectance is the amount of light energy reflected from the crystal. Expression for reflectance \((R)\) is given below:

\[
R = 1 \pm \sqrt{1 - \exp(-\alpha t) + \exp(\alpha t)} \frac{1}{1 + \exp(-\alpha t)},
\]

where \(\alpha\) is the absorption coefficient and \(t\) – the thickness of the sample [17,18].

The variation of reflectance with wavelength is depicted in Figure 8. The result indicates that the reflectance of the crystal is low.

![Figure 7. Plot of extinction coefficient versus wavelength for MSLO crystal.](image)

![Figure 8. Reflectance spectrum of MSLO crystal.](image)
3.2.5 Linear refractive index

Linear refractive index is the ratio of speed of light in vacuum or free space to the speed of light in the medium and it is determined using the following expression:

\[ n = \frac{1 + \sqrt{R}}{1 - \sqrt{R}} , \]

where \( R \) is the reflectance in decimal point. Plot of linear refractive index with photon energy for MSLO crystal is presented in Figure 9. At high wavelength region, the refractive index is observed to be increasing with increase of photon energy and it is very high near the fundamental absorption region.

![Plot of refractive index versus photon energy for MSLO crystal](image)

Figure 9. Variation of refractive index with photon energy for MSLO crystal.

3.2.6 Optical conductivity

The optical conductivity of the grown crystal was determined using the following relation:

\[ \sigma_{op} = \varepsilon_0 c n \alpha , \]

where \( c \) is the velocity of light in free space, \( \alpha \) is the linear absorption coefficient, \( \varepsilon_0 \) is the permittivity of the free space or vacuum and \( n \) is the refractive index. As per the relation above, optical conductivity is directly proportional to absorption coefficient and hence a straight line plot is obtained in Figure 10. It is seen that optical conductivity increases as the absorption coefficient increases. The value of optical conductivity of the sample can also be calculated in Gaussian units by...
Figure 10. Plot of optical conductivity (in SI units) versus absorption coefficient for MSLO crystal.

Figure 11. Plot of optical conductivity (in Gaussian units) versus absorption coefficient for MSLO crystal.

using the following relation:

\[
\sigma_{\text{op}} = \frac{1}{4\pi\varepsilon_0} \varepsilon_0 \alpha, \\
\sigma_{\text{op}} = 9 \times 10^9 \varepsilon_0 \alpha.
\]

The above equations have been used [19] to determine the value of optical conductivity. It is shown in Figure 11.

3.3 Impedance analysis

Impedance is a parameter opposition to the AC current in a sample. The complex impedance is given by \( Z = Z' + jZ'' \), where \( Z' \) is the real part of impedance, \( Z'' \)
is the imaginary part of impedance and $j$ is the complex parameter. The measurement of impedance provides a lot of information such as grain boundary resistance, bulk resistance, DC conductivity, electrical relaxation phenomena, etc. Impedance measurement for MSLO crystal was carried out using an Jognic’s make 2816 B Model impedance analyser over the frequency range of $1 \text{ Hz} - 10^6 \text{ Hz}$ at $30^\circ C$ and $50^\circ C$. The frequency-dependent real part and imaginary part of impedance for MSLO crystal are given in Figures 12 and 13. It is noticed

![Image](image-url)

**Figure 12.** Real part of impedance variation with frequency at different Temperatures for MSLO crystal.

![Image](image-url)

**Figure 13.** Imaginary part of impedance variation with frequency at different Temperatures for MSLO crystal.

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Figure 14. Nyquist plot of the grown MSLO crystal.

from the figures that, as the frequency increases, both the real and imaginary components of impedance decreases and appears to converge at the higher frequency side. This may be due to reduction of space charge polarization. Further, as temperature increases, impedance values increase and this indicates that the sample has negative temperature coefficient.

Nyquist plot of the grown MSLO crystal for different temperatures is shown in Figure 14. The result reveals the electrical characteristics of MSLO crystal are mostly caused by grain boundary and bulk effects [20, 21].

3.4 Vickers microhardness study

Microhardness measurement was carried out using a Vickers Microhardness indenter. Vickers hardness test comes under static indentation test method and this method is used to study the Microhardness of the grown crystals. Static indentation test is the simplest test, in which, specific geometry is pressed into the surface of specimen with known load. Ball or diamond cone or diamond pyramid intended shall be used. After removal of indenter, a permanent impression will be retained in the specimen. Hardness of the sample shall be calculated from the depth of indentation produced by measuring the cross sectional area or the depth of the indentation. The Vickers hardness number ($H_v$) or Diamond Pyramid Number (DPN) is estimated using the relation

$$ H_v = \frac{1.8554P}{d^2}, $$

where $P$ is the applied load and $d$ is the average diagonal length of the indented impression and 1.8544 is a constant of a geometrical factor for the diamond pyramid indenter [22]. Variations of average diagonal indentation length ($d$)
with load \((P)\) for the samples are shown in Figure 15. It is seen that the values of \(d\) are increasing as the applied load increases. “\(d\)” value of MSLO crystal is observed to be less compared to that of pure L-leucinium oxalate crystal.

Plots of hardness with applied load for both samples is shown in Figure 16. Hardness is observed to be more for MSLO crystal compared to that of undoped L-leucinium oxalate crystal. Observed high hardness value for MSLO crystal
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is due to presence of magnesium sulfate in the form of ions in the lattice of L-leucinium oxalate crystal.

3.4.1 Work hardening coefficient

Meyer gave a relationship between hardness and work hardening capacity of a material. It is given by \( P = (a \times d^n) \), where \( a \) and \( n \) are constants for a given material. The value of \( n \) can be considered as a parameter representing the capacity for work hardening or Meyer’s index and it may be determined experimentally by performing the test at various loads. The above relation can be written as

\[
\log(P) = \log(a) + n \log(d)
\]

Figure 17. Plot of \( \log(P) \) versus \( \log(d) \) for undoped L-leucinium oxalate crystal.

Figure 18. Plot of \( \log(P) \) versus \( \log(d) \) for MSLO crystal.
\[ \log P = \log(a) + n \log(d) \] and the slope of the line plotted between \( \log(d) \) and \( \log(P) \) gives the value of \( n \). The plots of \( \log(P) \) versus \( \log(d) \) for undoped and magnesium sulfate-doped L-leucinium oxalate (MSLO) crystals are presented in Figures 17 and 18. From the figures, the values of work hardening coefficient are found to be 2.236 and 2.517 respectively for undoped and MSLO crystals. As these values are more than 1.6, the grown crystals are confirmed to be soft category of materials [23, 24].

### 3.4.2 Fracture toughness

Fracture toughness \((K_c)\) is the resistance to fracture. It shows the toughness of a material and also it shows how much fracture is created in the material. It can be calculated using the equation

\[ K_c = \frac{P\beta C^2}{2}, \]

where \( \beta \) is the indenter constant equal to 7 for Vickers indenter and \( C \) is the crack length measured from the center of the indentation mark on the sample. The calculated values of fracture toughness at 50 g are \( 6.728 \times 10^5 \) kg/m\(^{3/2} \) and \( 7.021 \times 10^5 \) kg/m\(^{3/2} \) respectively for undoped and magnesium sulfate-doped L-leucinium oxalate crystals.

### 3.4.3 Brittleness

Brittleness is an important parameter which disturbs the mechanical behavior of a material. Value of brittleness index \((B_i)\) can be found using the equation

\[ B_i = \frac{H_v}{K_c}. \]

Estimated values of brittleness index at 50 g are \( 7.16 \times 10^{-5} \) and \( 8.05 \times 10^{-5} \) respectively for undoped and magnesium sulfate-doped L-leucinium oxalate crystals. Since, the values of hardness, fracture toughness and brittleness index of

<table>
<thead>
<tr>
<th>Applied load (g)</th>
<th>Hardness (kg/mm(^2))</th>
<th>Yield strength (10^6) (Pa)</th>
<th>Stiffness constant (Pa)</th>
</tr>
</thead>
<tbody>
<tr>
<td>10</td>
<td>55.10</td>
<td>2.422459</td>
<td>1.91276E+15</td>
</tr>
<tr>
<td>20</td>
<td>101.9</td>
<td>4.480010</td>
<td>5.60984E+15</td>
</tr>
<tr>
<td>30</td>
<td>137.0</td>
<td>6.023174</td>
<td>9.41687E+15</td>
</tr>
<tr>
<td>50</td>
<td>195.3</td>
<td>8.586320</td>
<td>17.5136E+15</td>
</tr>
<tr>
<td>70</td>
<td>220.6</td>
<td>9.698629</td>
<td>21.6748E+15</td>
</tr>
<tr>
<td>80</td>
<td>205.4</td>
<td>9.030364</td>
<td>19.1292E+15</td>
</tr>
<tr>
<td>100</td>
<td>187.2</td>
<td>8.230205</td>
<td>16.2623E+15</td>
</tr>
</tbody>
</table>
MSLO crystal are large, the grown crystal shall be useful in fabrication of NLO devices. Values of hardness, yield strength and stiffness constant are given in Table 2.

3.5 SHG efficiency

Kurtz and the Perry powder technique [25] was adopted to determine the relative Second Harmonic Generation (SHG) efficiency of MSLO crystalline sample. Beam of Nd: YAG laser was focused on the powdered sample of the crystal. Wavelength 1064 nm, pulse width 8ns and repetition rate of 10 Hz was used. Emission of green radiation at 532 nm indicates the SHG property and the values in connection with this test are provided in Table 3. KDP is used as reference sample. It emits green laser radiation of output energy of 8.8 mJ/pulse. For comparison purpose, the SHG study was also carried out for undoped L-leucinium oxalate crystal. It emitted the green laser radiation of output energy of 7.81 mJ/pulse. The grown MSLO crystal emits the green laser radiation of output energy of 10.05 mJ/pulse and hence the relative SHG efficiency of MSLO crystal is 1.14 times that of KDP crystal. The reason for the increase in output energy of magnesium sulphate doped L-leucinium oxalate crystal is due to the presence of dopant magnesium sulfate. Since magnesium sulfate-doped L-leucinium oxalate (MSLO) crystal has more SHG efficiency than that undoped L-leucinium oxalate and potassium dihydrogen phosphate (KDP) crystals, the grown crystal of MSLO could be used for better NLO applications.

<table>
<thead>
<tr>
<th>Sl. No.</th>
<th>Sample code/ Name of the sample</th>
<th>Output energy (mJ/pulse)</th>
<th>Input energy (J/pulse)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>KDP (Reference)</td>
<td>8.80</td>
<td>0.70</td>
</tr>
<tr>
<td>2</td>
<td>Undoped L-leucinium oxalate</td>
<td>7.81</td>
<td>0.70</td>
</tr>
<tr>
<td>3</td>
<td>MSLO crystal</td>
<td>10.05</td>
<td>0.70</td>
</tr>
</tbody>
</table>

3.6 Laser damage threshold study

Laser damage threshold (LDT) damage in NLO crystals is usually caused by avalanche and multi-photon ionizations and temperature rise in the sample etc. and it depends on many factors like wavelength, energy, pulse duration, longitudinal and transverse modes and beam size [26]. Laser damage threshold (LDT) value of grown crystal was measured using an Nd: YAG laser of wavelength of 1064 nm and pulse width of 10 ns. The laser beam with 1 mm diameter was focused by 30 cm focal length lens. To vary the energy of laser pulses an attenuator was employed. Pulse energy of every shot was determined by combining an oscilloscope and a phototube. "The value of LDT was estimated using the formula \( E/(\pi\tau r^2) \), where \( E \) is the input energy in mJ/pulse, \( \tau \) is the pulse width
and \( r \) is the radius of the laser spot. The calculated value of LDT of MSLO crystal is 1.85 GW/cm\(^2\). Since LDT value of this crystal is observed to be quite a large value and hence the grown crystal of MSLO could be used as the better crystal in NLO and laser applications.

3.7 Energy Dispersive Spectral (EDS) analysis

Precise identification of the components in a crystal is an essential requirement for microanalysis. EDAX-INCA Instrument is applied to confirm the composite elements of the grown crystal. The recorded EDS spectrum of MSLO crystal is shown in Figure 19. From the figure, we can see that, the MSLO crystal has various elements such as Mg, S and C, N, O.

![Figure 19. EDS spectrum of MSLO crystal.](image)

4 Conclusion

Magnesium sulfate-doped L-leucinium oxalate (MSLO) single crystals were grown by slow evaporation technique at room temperature (30°C). Structural analysis was carried out by single-crystal X-ray diffraction study and the lattice parameters of MSLO crystal were determined. The data reveals that, MSLO crystal crystallizes in triclinic structure with space group P1. The optical transmittance study was done to find various linear optical parameters such as band gap, reflectance, refractive index, absorption coefficient, extinction coefficient, optical conductivity. The spectrum reveals that the grown MSLO crystal’s lower cut-off wavelength is 218 nm. The calculated value of band gap is 5.69 eV and this value is observed to be high. The extinction coefficient increases slightly with increase of wavelength. The reflectance of the sample is low. The refractive
index is observed to be increasing with increase of photon energy. Impedance studies were performed for the grown sample to understand the electrical properties. It is found that, as the frequency increases, both the real and imaginary components of impedance decreases and appears to converge at the higher frequency side. Further, as the temperature increases, the impedance values increase and this indicates that the sample has the negative temperature coefficient. Vickers Microhardness study was carried out to find the mechanical properties such as hardness, work hardening coefficient, brittleness index and fracture toughness of the grown MSLO crystal. It is seen that the values of average diagonal indentation length ($d$) are increasing as the applied load increases and value of $d$ of MSLO crystal is observed to be less compared to that of undoped L-leucinium oxalate crystal. Hardness is observed to be more for MSLO crystal. Work hardening coefficient is found to be 2.236. Kurtz-Perry powder technique with Nd: YAG laser was employed to study the second harmonic generation (SHG) efficiency. SHG efficiency of MSLO crystal is 1.14 times that of KDP crystal. Z-scan technique was used to fine the third-order NLO parameter of the sample. The nonlinear third-order optical susceptibility of MSLO crystal is higher than those of many other crystalline materials. The result indicates that the wavelength of the emitted light is greater than excitation wavelength. LDT study was also done for MSLO crystal. The calculated value of LDT of MSLO crystal is 1.85 GW/cm$^2$. Since LDT value of this crystal is observed to be quite a large value, the grown crystal of MSLO could be used as the better crystal in Laser applications. SEM and EDS studies were conducted on MSLO crystal to identify elements present in the sample. From EDS Spectrum, it is noted that, the grown MSLO crystal contains the elements like Mg, S and C, N, O.

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Conflicts of Interest

The authors declare no conflict of interest.

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